

Na<sup>+</sup> Q<sup>-</sup> Na<sup>+</sup>



# **Sodium Sulfate**

## **Product Information**

Chemical name :	Sodium Sulfate, ACS Syn.: Na2SO4 , sodium	S grade salt of sulfuric acid, Disodiur	n sulfate; Bisodium sulfate;	0=\$=0
Dibasic sodium sulfate;	Disodium monosulfate; (All.): disodium sulphate; Natriumsulfat; sodium sulphate			
	CAS: 7757-82-6	<b>EC numbe</b> r: 231-82	20-9	
Cat. Number :	08762A, 500g	08762B, 1Kg	08762C, 2.5Kg	08762-B, Bulk
Properties:	Structure : Na2SO4 (anh.)Molecular Weight : 142.04Form: white solid crystalline powder (hygroscopic)Density: 2.664 g/cm³Melting point: 884°CBoiling point: 1429°CRefractive index (nD)/ 1.468pKa: 10.329 and 6.351(carbonic acid)Solubility: readily soluble in water >4% w/v			
Storage:	Room temperature $_{(Z)}$			
Safety:	Irritant. Non-flammable.		Risk Statements: 36. Safety Statements: 3	

Applications: Suitable for many biochemistry and biotechnology applications.





#### FT-08762A

Test	Specifications	
Purity	>99.0%	
Calcium (%)	0.01	
Chloride (%)	0.001	
Heavy Metals (as Pb)	0.0005	
Iron (%)	0.001	
Magnesium (%)	0.005%	
Nitrogen compounds (%)	0.0005%	
Phosphates (%)	0.001%	
Insolubles	0.001%	
Loss on Ignition (%)	0.5	
pH (5%, water, 25°C)	0.01	

#### **Specifications**

### **Technical information**

Chemistry

Sodium sulfate is a neutral salt, which forms aqueous solutions with pH of 7. The  $Na^+$  ion weakly polarizes its water ligands provided there are metal ions in solution.

Sodium sulfate reacts with sulfuric acid to give the acid salt sodium bisulfate, leading ot a temperatude-dependant equilibrium:

 $Na_2SO_4 + H_2SO_4 \rightleftharpoons 2 NaHSO_4$ 

Sulfate ions  $(SO_4^{2-})$  in solution can be indicated by the easy formation of insoluble sulfates when these solutions are treated with  $Ba^{2+}$  or  $Pb^{2+}$  salts:

 $Na_2SO_4 + BaCl_2 \rightarrow 2 \ NaCl + BaSO_4$ 

Double salts with some other alkali metal sulfates are known, including Na<sub>2</sub>SO<sub>4</sub>·3K<sub>2</sub>SO<sub>4</sub>.

• Solubility in water is high: 4.76 g/100 mL (0 °C), 42.7 g/100 mL (100 °C).

It is soluble in glycerol and hydrogen iodide, but insoluble in ethanol.

• Safety

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Although sodium sulfate is generally regarded as non-toxic, it should be handled with care. The dust can cause temporary asthma or eye irritation; this risk can be prevented by using eye protection and a paper mask. Transport is not limited

#### **Ordering information**

Catalog size quantities and prices may be found at <u>http://www.interchim.com</u>. Please inquire for higher quantities (availability, shipment conditions).

For any information, please ask : Uptima / Interchim; Hotline : +33(0)4 70 03 73 06

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