



# **DMNP-EDTA, NP-EGTA**

*Non-Fluorescent Caged Ca<sup>2+</sup> photolabile chelator* 

# **Product Information**

Name: DMNP-EDTA, AM ester Cell permeable

1-(4,5-dimethoxy-2-nitrophenyl)-1,2-diaminoethane-N,N,N',N'-tetraacetic acid,

tetra(acetoxymethyl ester); DMNPE, BNPE

Also called called DM-Nitrophen., a trademark of Calbiochem-Novabiochem Corp.

NO<sub>2</sub>

CHN(CH 2- C-OCH 20-C-CH 2)2

Kd for Ca<sup>2+</sup> increases from 5 nM to 3 mM upon photolysis.

**Catalog Number:** FP –M1437A, 1 mg

**Molecular Weight:** 761.66

**Specifications :** Appearance: Yellow oil HPLC Purity: >95%

TLC (in 1:1 Hexane/Ethyl acetate): conform NMR (CDCl<sub>3</sub>): conform to reference structure

Soluble in: DMSO, Methanol, DMF, CH<sub>3</sub>CN and CHCl<sub>3</sub>

**Excitation/Emission**:  $\lambda_{\text{exc}}/\lambda_{\text{em}}$  (EtOH) = 332 nm/ N/A

**Extinction Coefficient:** 4 550 M<sup>-1</sup> cm<sup>-1</sup>

Storage: -20°C (M). Protected from light and moisture.

Name: DMNP-EDTA Cell impermeable

**Catalog Number:** FP –44506A, 5 mg

Molecular Weight: 473,4

Soluble in: DMSO, Methanol, DMF and basic water

Storage: -20°C (M). Protected from light and moisture.



FT-M1473A

Name: DMNP-EDTA, tetrapotassium salt (caged calcium) Cell impermeable

Catalog Number: FP –OO5460, 5 mg

**Molecular Weight:** 626

Storage: -20°C (M). Protected from light and moisture.

Name: NP-EGTA, AM ester \*Cell permeable\*

O-NitroPhenyl EGTA TetraPotassium Salt.

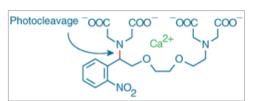
Catalog Number: FP –M1433A, 1 mg

Also available as K salt # 52902A

 $\label{eq:C32} \textbf{Structure:} \qquad \qquad C_{32}H_{43}N_3O_{20}$ 

Molecular Weight: 789.7

**Soluble in :** DMSO, Methanol, DMF, CH<sub>3</sub>CN and CHCl<sub>3</sub>



## Introduction

#### DMNP-EDTA

1-(4,5-dimethoxy-2-nitrophenyl)ethyl EDTA (DMNPE) is a caging group with long-wavelength absorption (absorption maximum  $\sim$ 355 nm), developed by Ellis-Davies and Kaplan Kaplan Morad 1988. Morad 1988. It absorbs 340–360 nm light very efficiently. We provide the salted acid form, and the AM ester to ease cell loading.

DMNP-EDTA is used to raise or lower the intracellular  $Ca^{2^+}$  concentration in cells, in order to study the resulting effects. It also can be used as a an effective caged  $Mg^{2^+}$  source, with even faster photorelease of  $Mg^{2^+}$ , and furthermore for photolytic release of other divalent cations such as  $Sr^{2^+}$ ,  $Ba^{2^+}$ ,  $Mn^{2^+}$ ,  $Co^{2^+}$  and  $Cd^{2^+}$ .

Upon illumination, DMNP-EDTA's is cleaved and dissociation constant ( $K_d$ ) for  $Ca^{2^+}$  increases from 5 nM to 3mM for the two iminodiacetic acid photolysis products ( $\sim$ 600,000-fold lower than that of DMNP-EDTA). Using DMNP-EDTA complexed to  $Ca2^+$ , this results in a pulse of free  $Ca^{2^+}$ . Using the AM form,  $Ca2^+$  can be scavenged in cells upon photolysis. Additionally, DMNP-EDTA has significantly higher affinity for  $Mg^{2^+}$  ( $K_d = 2.5~\mu M$ ) than does NP-EGTA ( $K_d = 9~mM$ ), increasing after photolysis to  $K_d$  for  $Mg^{2^+}$  to  $\sim$ 3 mM . Photorelease of  $Ca^{2^+}$  is rapid, occuring in <180 microseconds, with even faster photorelease of  $Mg^{2^+}$ . EDTA prevents the potentially cytotoxic reaction between amines and the 2-nitrosobenzoyl photolytic by-products.

See use and limitation of DMNP-EDTA in the literature (reviews Ellis-Davies 2007, 2008).

## **Literature – DMNP-EDTA**

- **Bollmann JH** *et al.* "Control of synaptic strength and timing by the release-site Ca2+ signal.", Nat Neurosci 8, 426-34 (2005) PN56193. [DMNP-EDTA] <u>PubMed</u> ·
- Ellis-Davies GC. Caged compounds: photorelease technology for control of cellular chemistry and physiology. Nat Methods (2007) 4:619-628
- Ellis-Davies GC. Neurobiology with caged calcium. Chem Rev (2008) 108:1603-1613 (Review of neuroscience applications of photoactivatable ("caged") calcium reagents)
- Faas GC et al. "Kinetic properties of DM-nitrophen binding to calcium and magnesium.", Biophys J 88, 4421-33 (2005) PN57135. [DMNP-EDTA] PubMed · Article ·
- Girouard H. et al., Astrocytic endfoot Ca2+ and BK channels determine both arteriolar dilation and constriction, PNAS, 2010; 10.1073/pnas. 0914722107 <u>Abstract</u>



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- Gusev K. et al., Angiotensin II—Mediated Adaptive and Maladaptive Remodeling of Cardiomyocyte Excitation—Contraction Coupling, Circ. Res., 105: 42 - 50 (2009) <u>Article</u>
- Kaplan JH, Ellis-Davies GC, Photolabile chelators for the rapid photorelease of divalent cations. *Proc Natl Acad Sci* U S A (1988) 85:6571-6575
- Morad M, Davies NW, Kaplan JH, Lux HD. Inactivation and block of calcium channels by photo-released Ca2+ in dorsal root ganglion neurons. *Science* (1988) 241:842-844

#### NP-EGTA AM

NP-EGTA AM is loaded in cells where intracellular esterases cleaves the AM group, preventing release from cells. It acts then as a chelator with a high selectivity for  $Ca^{2+}$ . Upon UV illumination, its Kd raises from 80 nM to >1 mM (affinity decrease by 12 500-fold). It as also a high photochemical quantum yield ref (~0.2). Furthermore, NP-caged EGTA does not perturb physiological levels of  $Mg^{2+}$  (Kd for  $Mg^{2+}$  is 9 mM).

The NP-EGTA salt can be complexed with  $Ca^{2+}$  to generate a caged calcium complex that will rapidly deliver  $Ca^{2+}$  upon photolysis. The cell-permeant AM ester of NP-EGTA does not bind  $Ca^{2+}$  unless the AM esters are removed. It can potentially serve as a photolabile buffer in cells because, once converted to NP-EGTA by intracellular esterases, it will bind  $Ca^{2+}$  with high affinity until photolyzed with UV light. NP-EGTA has been used to measure the calcium buffering capacity of cells.

# **Related products**

- Fluo-3 AM, FP-78932C
- Fluo-8 AM, CP7502
- TPEN (Tetrakis-(2-pyridylmethyl)ethylenediamine), FP-44736A
- Pluronic acid, FP-37361A
- Caged Ca: NP-EGTA, FP-52902A

- DMNP-EGTA, <u>FP-44506A</u> and –AM <u>FP-M1437A</u>
- Ionomycin, FP-53989A
- Fluo-3 AM, FP-78932A
- Fluo-8 NW, CJ2560
- Calcium Calibration Kit, FP-21527A
- Trivalent lanthanide terbium (III), a luminescent analog of Ca<sup>2+</sup> that can be used to study structure—function relationships in Ca<sup>2+</sup>-binding proteins such as calmodulin, oncomodulin, lactalbumin and ATPases.
- Thapsigargin, a sesquiterpene lactone that promote releases Ca<sup>2+</sup> from intracellular stores but does not directly affect plasma membrane Ca<sup>2+</sup>-ATPases, Ins 1,4,5-P3 production or protein kinase C activity.

# **Ordering information**

Catalog size quantities and prices may be found at <a href="http://www.interchim.com">http://www.interchim.com</a>. Please inquire for higher quantities (availability, shipment conditions).

For any information, please ask: FluoProbes<sup>®</sup> / Interchim; Hotline: +33(0)4 70 03 73 06

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